

# SCR Impact on Mercury Speciation in Coal-Fired Boilers

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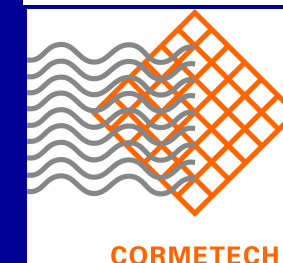
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# Background

- Speciation influences emissions control
  - Ionic  $\text{Hg}^{2+}$  is removed easily by wet scrubbers
  - Volatile elemental  $\text{Hg}^0$  is difficult to capture
- Many Selective Catalytic Reduction (SCR) units are meeting stringent  $\text{NO}_x$  regulations
  - Vanadia/titania ( $\text{V}_2\text{O}_5/\text{TiO}_2$ ) catalyst
  - Ammonia ( $\text{NH}_3$ ) or Urea ( $\text{CH}_4\text{ON}_2$ ) reductant
- SCR has an impact on mercury speciation
  - Limited field data in Europe and U.S.
  - Increase in  $\text{Hg}^{2+}$  across SCR reactor

# Factors Affecting Hg Chemistry

- Apparent dependence on coal type
  - Higher  $\text{Hg}^{2+}$  across SCR for bituminous coal-fired boilers
  - Little change in Hg speciation across SCR for boilers burning sub-bituminous (Powder River Basin) coal
- Possible effects of SCR system
  - Changes in flue gas chemistry ( $\text{NO}_x$ ,  $\text{NH}_3$ ,  $\text{Cl}_2$ ,  $\text{SO}_3$ )
  - Catalytic oxidation by vanadium based catalysts
- Important reactions transforming  $\text{Hg}^0$  to  $\text{Hg}^{2+}$  in SCR systems are not well understood

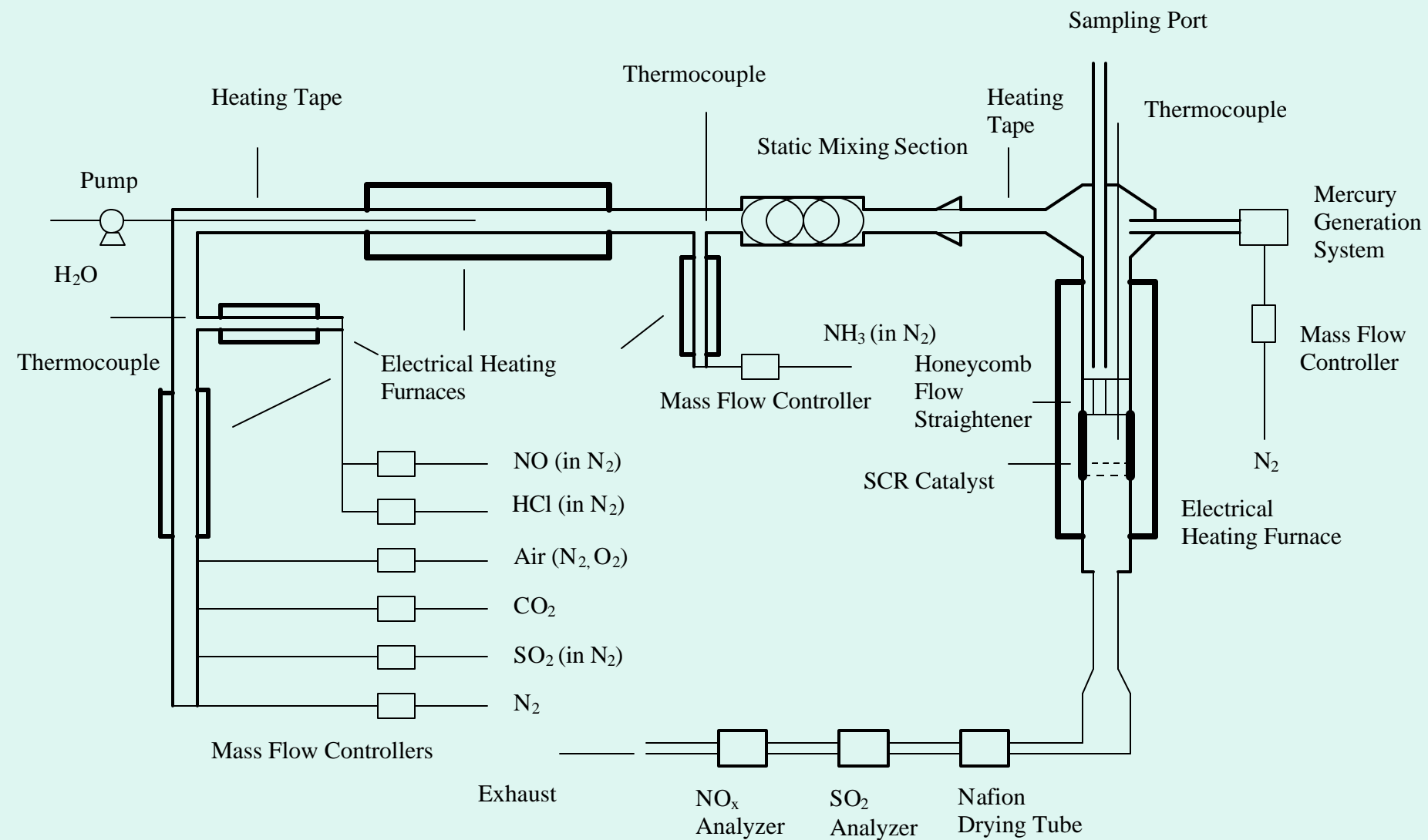
# Objectives

- Identify important parameters for enhancing  $\text{Hg}^0$  oxidation in SCR systems
- Provide scientific base for apparent coal-type dependence on SCR effect on  $\text{Hg}^0$  oxidation
- Better understanding of the fundamentals of SCR enhanced mercury oxidation for developing multi-pollutant control strategies

# Approach

- Good control on experimental variables
  - Bench-scale SCR reactor
  - Simulated combustion flue gases for bituminous and sub-bituminous coals
- Modified Ontario Hydro (OH) method for speciation sampling/analysis
  - Lower sampling volume
  - Mini-impingers

# SCR Reactor System

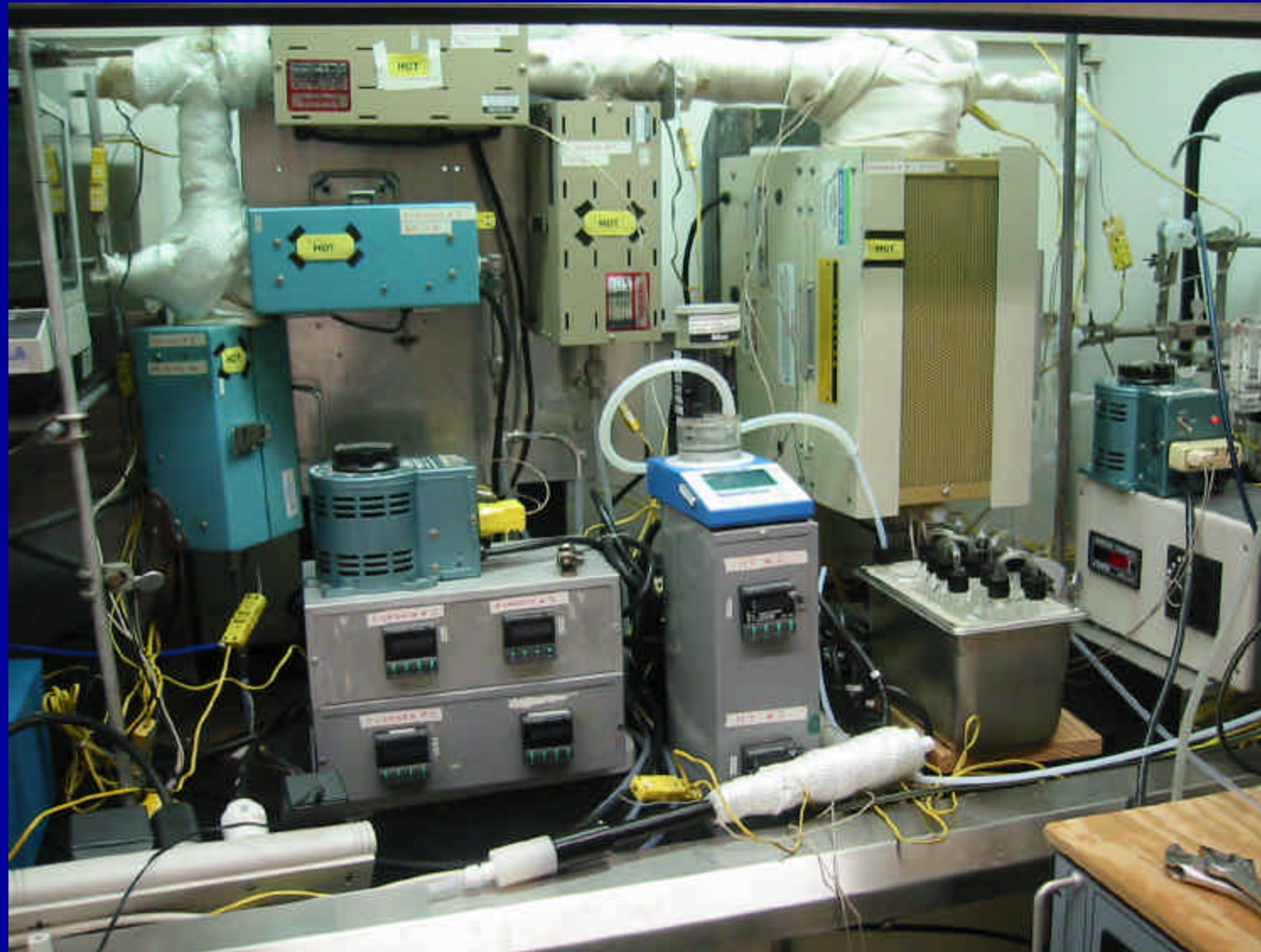


# Bench-Scale SCR Reactor System



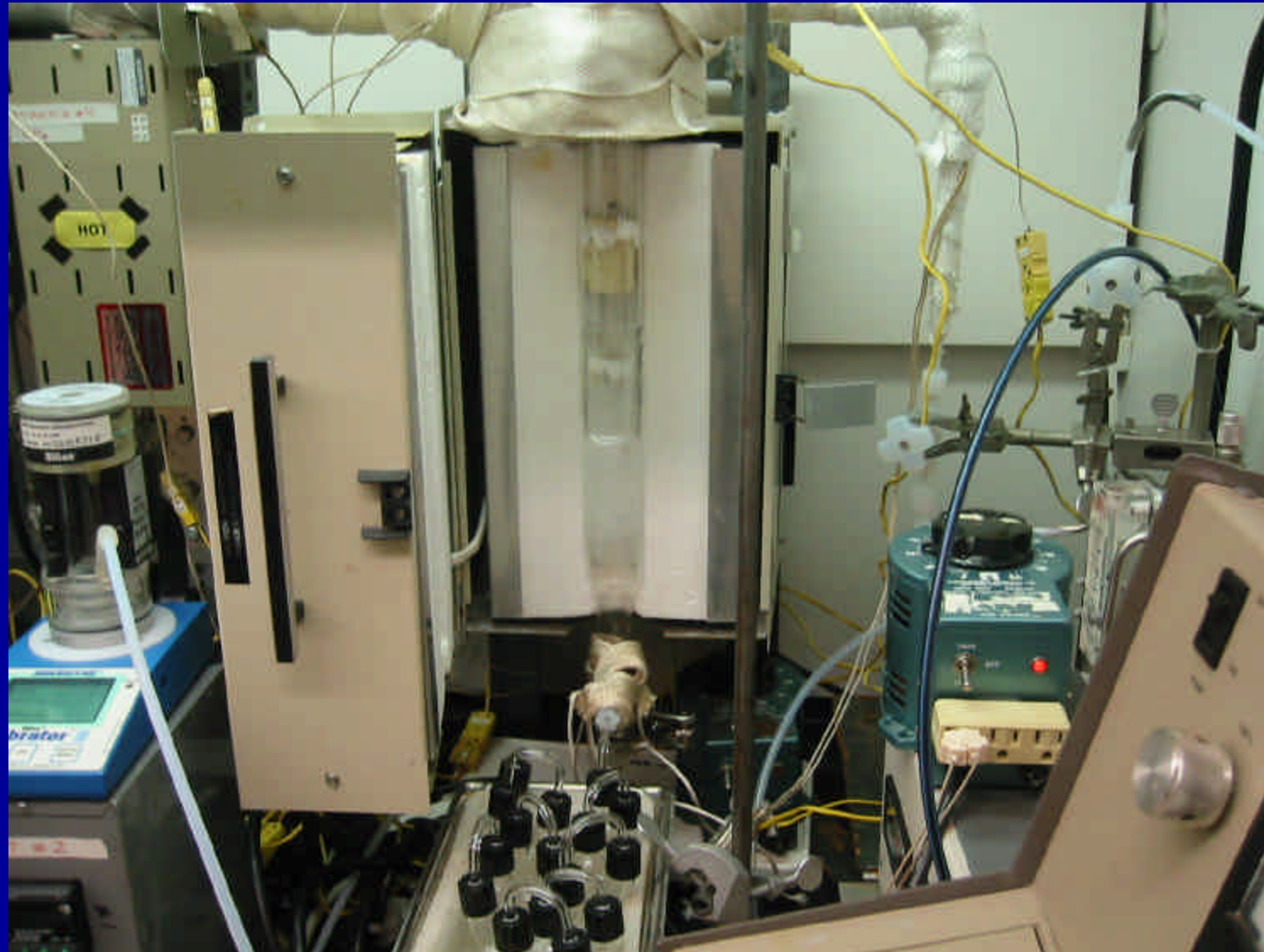


# Simulated Flue Gas Preheating and Mixing

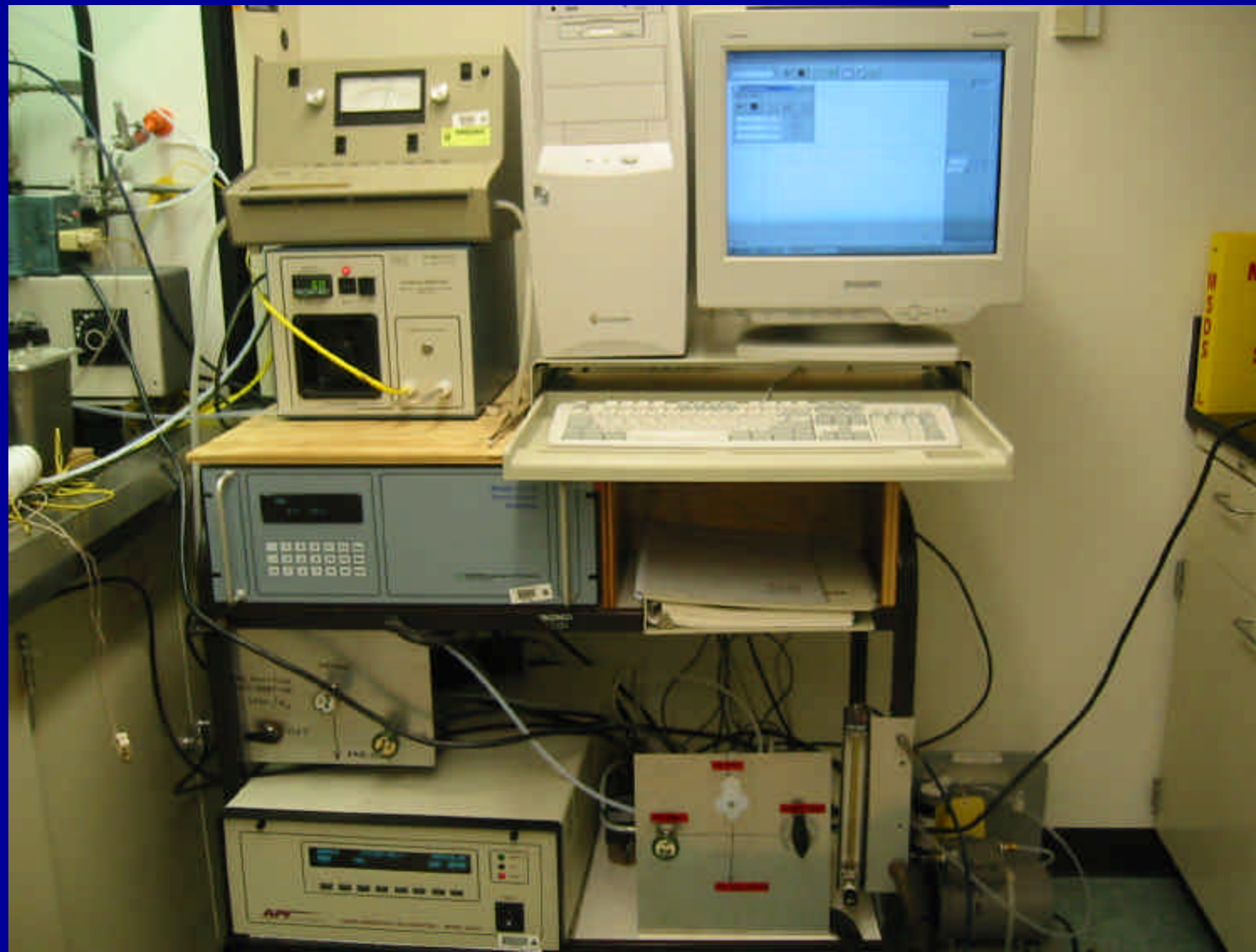




# SCR Reactor



# $\text{NO}_x$ and $\text{SO}_2$ Monitors



# Experimental Procedures

- Catalyst
  - Cormetech commercial honeycomb catalyst (2.2 x 2.2 x 1.9 cm, 9 channels)
  - Space velocity 2609 hr<sup>-1</sup> at 400 cm<sup>3</sup>/min gas flow rate
- Thermal pre-treatment of catalyst
  - Heating of catalyst overnight at 425 °C under N<sub>2</sub> flow
  - Minimize residual effect from previous experiment
- Catalyst pre-conditioning
  - Passing SO<sub>2</sub> and HCl through catalyst overnight at levels for next day's experiment
- Add remaining flue gas components (O<sub>2</sub>, CO<sub>2</sub>, H<sub>2</sub>O, NO, NH<sub>3</sub>, Hg<sup>0</sup>) before experiment

# Mercury Sampling and Analysis

- OH sampling
  - Sampling started after  $\text{NO}_x$  reached steady state level
  - Two hour sampling time ( $0.05 \text{ m}^3$  total sampling volume)
  - Measure sampling flow ( $400 \text{ cm}^3/\text{min}$ ) every 10 min
- Sampling impingers
  - $\text{Hg}^{2+}$  collected by first three impingers containing  $\text{KCl}$  (1N) solution
  - $\text{Hg}^0$  collected by one impinger containing  $\text{HNO}_3$  (5%) and  $\text{H}_2\text{O}_2$  (10%) solution and three impingers containing  $\text{H}_2\text{SO}_4$  (10%) and  $\text{KMnO}_4$  (4%) solution
- Each fraction prepared/analyzed for mercury by cold vapor atomic absorption (CVAA)

# Simulated Powder River Basin Coal Combustion Flue Gas Mixtures

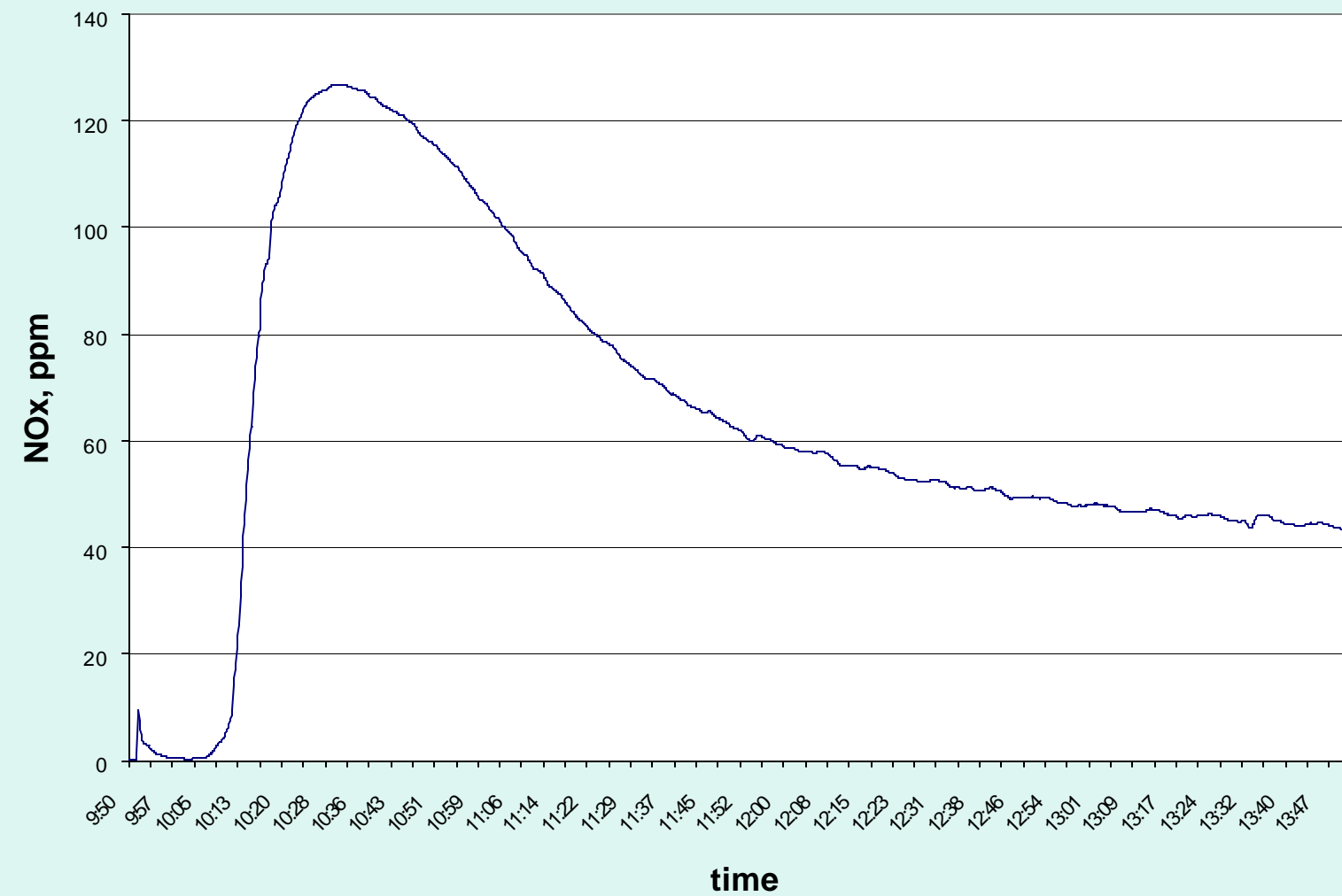
Test No.	P1	P2	P3	P4
Simulation	PRB coal	PRB coal without HCl	PRB coal without NH <sub>3</sub>	PRB coal without NH <sub>3</sub> and NO <sub>x</sub>
HCl Concentration (ppm)	8	0	8	8
SO <sub>2</sub> Concentration (ppm)	280	280	280	280
NO <sub>x</sub> Concentration (ppm)	350	350	350	0
NH <sub>3</sub> Concentration (ppm)	315	315	0	0
CO <sub>2</sub> Concentration (%)	15	15	15	15
O <sub>2</sub> Concentration (%)	3.5	3.5	3.5	3.5
H <sub>2</sub> O Concentration (%)	5.3	5.3	5.3	5.3
Hg <sup>0</sup> concentration (ppb)	19	19	19	19

# Simulated Bituminous Coal Combustion Flue Gas Mixtures

Test No.	B1	B2	B3	B4
Simulation	High Cl, low S	Medium Cl and S	B2 without SO <sub>2</sub>	Low Cl, high S
HCl Concentration (ppm)	204	134	134	98
SO <sub>2</sub> Concentration (ppm)	934	2891	0	3116
NO <sub>x</sub> Concentration (ppm)	350	350	350	350
NH <sub>3</sub> Concentration (ppm)	315	315	315	315
CO <sub>2</sub> Concentration (%)	15	15	15	15
O <sub>2</sub> Concentration (%)	3.5	3.5	3.5	3.5
H <sub>2</sub> O Concentration (%)	5.3	5.3	5.3	5.3
Hg <sup>0</sup> concentration (ppb)	19	19	19	19



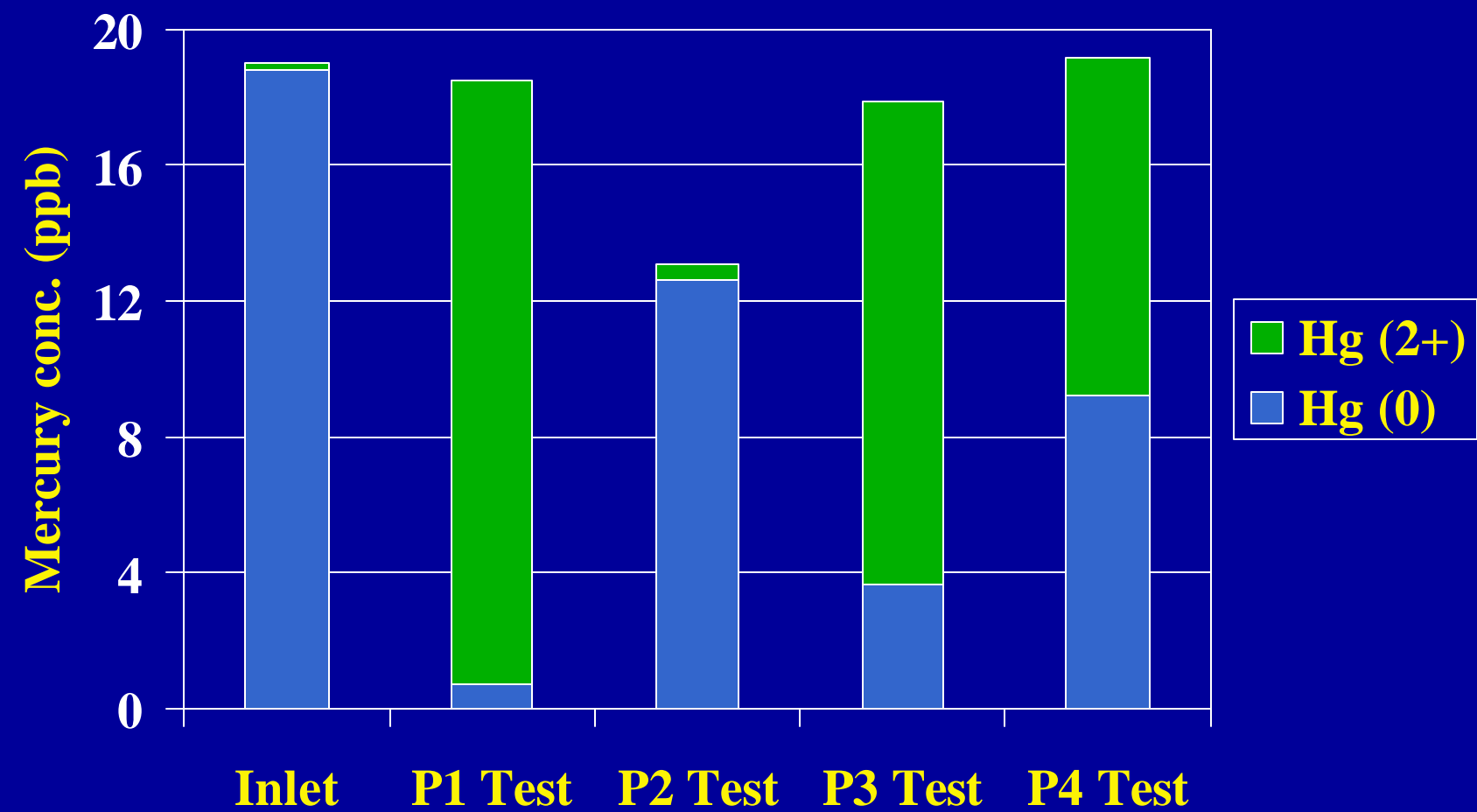
# SCR Outlet NO<sub>x</sub> Concentration Profile



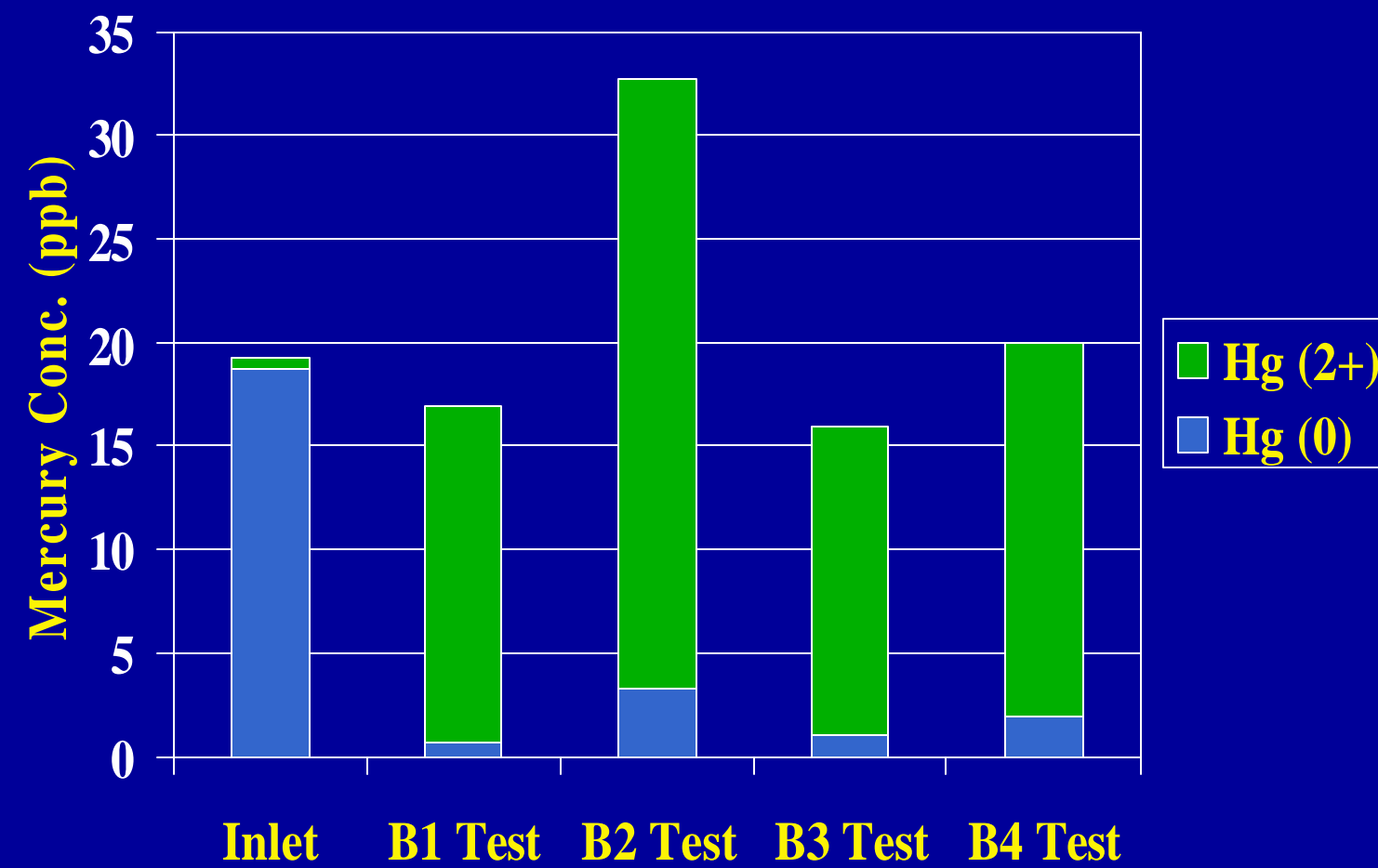
# NO<sub>x</sub> Reduction Results

Test No.	P1	P2	P3	P4	B1	B2	B3	B4
SCR Outlet NO <sub>x</sub> Concentration (ppm)	44	52	350	0	44	43	47	46
NO <sub>x</sub> Reduction (%)	87	85	0	0	87	88	87	87

## SCR Outlet Mercury Speciation Results for PRB Coal Simulation Experiments



## SCR Outlet Mercury Speciation Results for Bituminous Coal Simulation Experiments



# Discussion

- Chlorine is critical for  $\text{Hg}^0$  oxidation in SCR
  - Low Cl and high Ca in PRB coals cause little SCR impact
  - Cl in bituminous coals sufficient to cause significant impact
- Possible mechanisms involved over SCR catalyst
  - SCR catalyzed Deacon reaction:  $2\text{HCl} + 1/2 \text{O}_2 = \text{Cl}_2 + \text{H}_2\text{O}$
  - Chlorination reaction:  $\text{V}_2\text{O}_5 + 2\text{HCl} = 2\text{VO}_2\text{Cl(s)} + \text{H}_2\text{O}$
- $\text{NO}_x$  promotes  $\text{Hg}^0$  oxidation in SCR
  - $\text{NO}_x$  seems to play a significant role for  $\text{Hg}^0$  oxidation in SCR
  - Chemisorption of  $\text{NO}_x$  creates active sites for  $\text{Hg}^0$  adsorption
  - Reactions of  $\text{NH}_3$  with  $\text{NO}_x$  inhibit  $\text{Hg}^0$  adsorption
- $\text{SO}_x$  does not seem to play a significant role in SCR  $\text{Hg}^0$  oxidation under conditions tested to date
  - Suggests that  $\text{Hg}^0$  is unlikely to be oxidized by  $\text{SO}_3$

# Summary and Conclusions

- Bench-scale system simulated field units closely
  - Achieved NO<sub>x</sub> reduction levels similar to those in field units
- Different effects of flue gases on SCR Hg<sup>0</sup> oxidation
  - HCl provides critical chlorine source for Hg<sup>0</sup> oxidation
  - NO<sub>x</sub> has a significant promotional effect
  - SO<sub>x</sub> has little effect under the conditions of this study
- Complex interactions between Hg<sup>0</sup>, flue gases, and SCR catalyst result in Hg<sup>0</sup> oxidation
- Results provide scientific evidence for apparent coal-type effect on Hg<sup>0</sup> oxidation in SCR systems



# Future Work

- Effect of catalyst age
  - Aged samples collected in the field
- Effect of catalyst formulation
  - Catalysts for PRB coal application
- Effect of residence time
- Effect of  $\text{NH}_3/\text{NO}_x$  molar ratio
- Mechanistic and modeling studies

# Acknowledgement

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